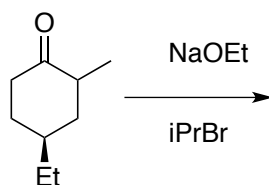
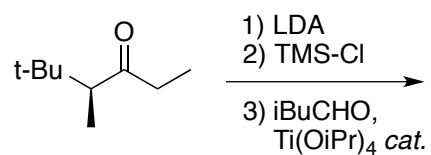
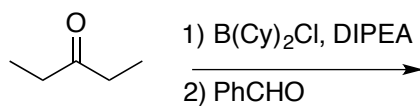


Chem 200
Fall 2014
Problem set 5
Due Monday Dec 15/2014

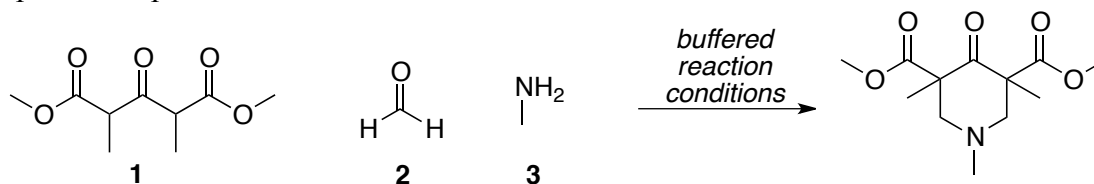
(6 Points) Draw the transition state or intermediate that is responsible for the observed regio- chemo- or stereoselectivity in each reaction. Draw the product you expect to form, including all relevant stereochemistry.



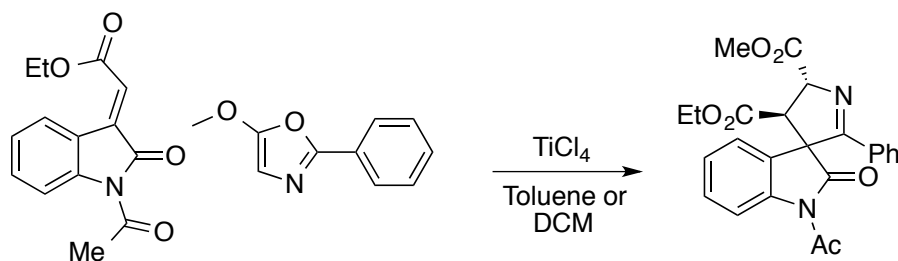
2) (4 Points)

When ketone **1** was mixed with excess formaldehyde (**2**) and methylamine (**3**) a cyclic amine was formed. Propose a mechanism that explains this reaction.

NOTE: The reaction is performed under buffered conditions so both basic and acidic species are present in the reaction media.



- 3) (4 Points) The Aldol reaction (enolate attacking an aldehyde or ketone) is the basis for many reactions that differ only slightly (imine in place of a carbonyl for example). Below is one such reaction, which is similar to a Mukaiyama aldol reaction. It only works when a Titanium catalyst is added and give the rearranged product shown.



- What is the mechanism of this reaction?
- How is this like an aldol reaction? What steps are in common and what are different?
- Why do you get only the trans product? Show an intermediate and make an argument that supports your answer.
- This reaction only works in non-coordinating solvents such as DCM or toluene. Why does the reaction fail in THF or MeOH?

- 4) (4 Points) A recent article was discussing the reaction of azomethine imines to make new compounds using a formal dipolar cycloaddition strategy.
- a. Find this paper and report the literature reference including:
 - i. Authors
 - ii. Journal
 - iii. Publication date
 - iv. Pages (if assigned)
 - v. DOI
 - b. Choose any compound in Figure 5 and draw out the starting materials you would use and the mechanism of the reaction
 - c. When the reaction uses a chiral Lewis base the reaction can be done enantioselectively. Using the models they present in the paper and our latest discussion explain how the Lewis base makes the reaction enantioselective